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# Pyridinium dihydrogenmonothiophosphate

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By reaction of pyridine dithiomonometaphosphoryl fluoride with ethyldiphenylphosphine in the presence of traces of water, we isolated crystals of the title compound, (I). In the structure of this salt, two kinds of hydrogen bonds were detected. They link N-H (part of the cation), as well as one O-H group (part of the anion), to the phosphoryl O atom of the anion. The donor-acceptor distances are 2.654 (1) and 2.568 (1) Å, respectively. The P-O and P-S bond lengths are 1.514 (2) and 1.967 (1) Å, respectively, which fall therefore in the range of a 1.5-fold bond.



### **Experimental**

According to Fluck et al. (1973), a solution of ethyldiphenylphosphine in benzene was slowly dropped into a suspension of pyridine dithiomonometaphosphoryl fluoride in benzene with stirring. After 1 h, the solid product was filtered and recrystallized from a solution in benzene/acetonitrile.

Crystal data

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C_5H_6N^+ \cdot H_2PSO_3^-
                                                        D_x = 1.612 \text{ Mg m}^{-3}
M_r = 193.15
                                                        Mo K\alpha radiation
Monoclinic, P2_1/n
                                                        Cell parameters from 1775
a = 6.7837 (13) \text{ Å}
                                                            reflections
b = 9.0268 (11) \text{ Å}
                                                        \theta = 2.5 - 25.0^{\circ}
                                                        \mu = 0.564 \text{ mm}^{-1}
c = 13.237 (2) \text{ Å}
\beta = 101.02 \ (2)^{\circ}
                                                         T = 180 (2) \text{ K}
V = 795.7 (2) Å<sup>3</sup>
                                                        Plate, vellow
Z = 4
                                                        0.40\,\times\,0.28\,\times\,0.12 mm
Data collection
```

Stoe IPDS diffractometer 1260 reflections with  $I > 2\sigma(I)$  $\varphi$ -oscill.,  $\varphi$ -incr. = 1.5°, 133 exposure  $R_{\rm int} = 0.038$  $\theta_{\rm max} = 26.04^\circ$ scans Absorption correction: see below;  $h=-8\rightarrow 8$  $T_{\min} = 0.806, \ T_{\max} = 0.935$  $k = -11 \rightarrow 10$ 4980 measured reflections  $l = -16 \rightarrow 16$ 1540 independent reflections

### Refinement

Refinement on $F^2$	All H-atom parameters refined
$R[F^2 > 2\sigma(F^2)] = 0.029$	$w = 1/[\sigma^2(F_o^2) + (0.0472P)^2]$
$wR(F^2) = 0.077$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.041	$(\Delta/\sigma)_{\rm max} = 0.001$
1538 reflections	$\Delta \rho_{\rm max} = 0.34 \ {\rm e} \ {\rm \AA}^{-3}$
126 parameters	$\Delta \rho_{\rm min} = -0.35 \ \rm e \ \rm \AA^{-3}$

For the studied crystal, a  $\Delta F^2$ -based absorption correction was carried out. The calculation made with ABSCOR (Stoe & Cie, 1997), a modification of DIFABS (Walker & Stuart, 1983). Unlike DIFABS. ABSCOR works with  $F^2$  values instead of F. All H atoms were refined [C-H 0.89 (2)-0.91 (2) Å].

Data collection: IPDS-2.87 (Stoe & Cie, 1997); cell refinement: IPDS-2.87 (Stoe & Cie, 1997); data reduction: IPDS-2.87 (Stoe & Cie, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XSTEP-2.18 (Stoe & Cie, 1997); software used to prepare material for publication: SHELXL97 (Sheldrick, 1997).

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